

## SAFETY DATA SHEET

This SDS adheres to the standards and regulatory requirements of the United States and may not meet the regulatory requirements in other countries.

1. Identification

Product identifier Sodium Chlorite Solution 26%-40%
Other means of identification Sodium Chlorite Solution, ERCOPURE

**Recommended use** Generation of chlorine dioxide for use as a disinfectant, or for use

as an oxidant. Bleaching of textiles and other fibers.

**Recommended restrictions** None known.

Manufacturer/Importer/Supplier/Distributor information

Manufacturer

**Company name** International Dioxcide, Inc. an ERCO Worldwide Company

Address 40 Whitecap Drive

North Kingstown, RI 02852 United States of America

TelephoneInformation #: (800) 477-6071Websitehttp://www.ercoworldwide.comE-mailproductinfo@ercoworldwide.com

Emergency phone number Canada & USA: 1-800-424-9300 (CHEMTREC)

**Supplier** Refer to Manufacturer

2. Hazard(s) Identification

Physical hazards None

Health hazards Acute toxicity, oral Category 4

Acute toxicity, inhalation Category 3
Serious eye damage Category 1
Specific target organ toxicity, repeated Category 2

exposure (blood, kidneys, liver, spleen)

Skin corrosive Category 1B

**Environmental hazards** Not currently regulated by OSHA, refer to Section 12 for additional

information.

OSHA defined hazards

This mixture does not meet the classification criteria according to OSHA

HazCom 2012.

**Label elements** 









Signal word Danger



**Hazard statement** Harmful if swallowed.

Toxic if inhaled.

Causes serious eye damage.

May cause damage to organs through prolonged or repeated exposure

(blood, kidneys, liver, spleen).

Causes severe skin burns and eye damage.

### **Precautionary statement**

**Prevention** Wear protective gloves, protective clothing, eye protection, face protection.

Do not eat, drink or smoke when using this product. Do not breathe dust, fume, gas, mists, vapors, spray. Wash hands and face thoroughly after

handling. Use only outdoors or in a well ventilated area.

**Response** IF SWALLOWED: Call a POISON CENTER or doctor, Rinse mouth. Do NOT

induce vomiting.

**IF ON SKIN (OR HAIR):** Take off immediately all contaminated clothing. Rinse skin with water (or shower). Wash contaminated clothing before reuse. **IF INHALED:** Remove person to fresh air and keep comfortable for breathing.

Call a POISON CENTER or doctor.

**IF IN EYES:** Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a

POISON CENTER or doctor.

If exposed or concerned: Call a POISON CENTER or doctor.

**Storage** Store in a well-ventilated place. Keep container tightly closed. Store locked up.

**Disposal** Dispose of contents and containers in accordance with

local/regional/national/international regulations.

Hazard(s) not otherwise classified (HNOC)

No OSHA defined hazard classes.

Other hazards which do not result in classification:

Contact with some metals will generate flammable hydrogen gas. Chronic skin contact with low concentrations may cause dermatitis.

Contact with acids or reducing agents will generate toxic chlorine dioxide gas. If Sodium Chlorite dries on some types of fire-retardant clothing it is known to cause an exothermic reaction. The reaction has been known to cause burns to skin. Nomex appears to be the only material not to experience this

reaction.

Supplemental information

Not applicable.

# 3. Composition/Information on Ingredients

Chemical name	Common name and synonyms	CAS number	Conc. % By Weight
Sodium Chlorite	None	7758-19-2	26-40 w/w%
Dihydrogen Oxide	Water	7732-18-5	Balance



Chemical name of impurities, stabilizing solvents and/or additives: None.

### 4. First-Aid Measures

**Inhalation** Remove person to fresh air and keep comfortable for breathing. If breathing is

difficult, trained personnel should give oxygen. If breathing stops, provide artificial

respiration. Immediately call a POISON CENTER or doctor/physician.

Skin Contact Take off immediately all contaminated clothing. Immediately flush skin with

running water for at least 20 minutes. Wash contaminated clothing promptly. Leather and shoes that have been contaminated with the solution may need to be

destroyed. Immediately call a POISON CENTER or doctor/physician.

**Eye Contact** Immediately flush eyes with plenty of water for at least 20 minutes. Remove

contact lenses, if present and easy to do so. Continue rinsing. Take care not to rinse contaminated water into the unaffected eye or onto the face. Immediately call a

POISON CENTER or doctor/physician.

Ingestion Rinse mouth. Do NOT induce vomiting. Never give anything by mouth to a victim

who is unconscious or is having convulsions. Call a POISON CENTER or

doctor/physician if you feel unwell.

Most important symptoms/effects, acute and delayed

Causes serious eye damage. Symptoms may include stinging, tearing, redness, swelling, and blurred vision. Permanent eye damage including blindness could result. May be harmful or fatal if swallowed. Symptoms may include pain, headache, nausea, vomiting, dizziness, drowsiness and other central nervous system effects. Can cause severe skin burns. Symptoms may include redness, edema, drying, defatting and cracking of the skin. Prolonged exposure may cause chronic effects. Material is irritating to mucus membranes and upper respiratory tract. Symptoms may include bloody nose and sneezing. High concentrations may cause lung damage.

Indication of immediate medical attention and special treatment needed

Immediate medical attention is required. Causes chemical burns. May be harmful or fatal if swallowed. Symptoms may be delayed.

General information

Ensure that medical personnel are aware of the material(s) involved and take precautions to protect themselves. Show this safety data sheet to the doctor in attendance.

## 5. Fire-Fighting Measures

Suitable extinguishing media

Water spray, fog (flooding amounts). Water only; no dry chemical,  $CO_2$  or Halon. This product itself does not burn but combustibles wetted with this solution and subsequently dried are easily ignited and burn vigorously.

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Unsuitable extinguishing media

DO NOT use dry chemical fire extinguishing agents containing ammonium compounds (such as some A:B:C agents), since an explosive compound can be formed. DO NOT use carbon dioxide, dry chemical powder or other extinguishing agents that smother flames, since they are not effective in extinguishing fires involving oxidizers. Use chemical extinguishing agents with caution.

Specific hazards arising from the chemical Special protective equipment and precautions for firefighters

May intensify fire; oxidizer when dry. Drying of this product on clothing or combustible materials may cause fire.

Firefighters must use standard protective equipment including flame retardant coat, helmet with face shield, gloves, rubber boots, and in enclosed spaces, SCBA.

If Sodium Chlorite dries on some types of fire-retardant clothing it is known to cause an exothermic reaction. The reaction has been known to cause burns to skin. Nomex appears to be the only material not to experience this reaction.

Firefighting equipment /instructions

Evacuate area. Remove all sources of ignition. In case of fire: Stop leak if safe to do so. Move combustibles out of path of advancing pool if you can do so without risk. Move containers from fire area if you can do so without risk. Fight fire from upwind to avoid exposure to combustion products. In case of fire and/or explosion do not breathe fumes.

Specific methods

Use standard firefighting procedures and consider the hazards of other involved materials.

**General fire hazards** 

May intensify fire; oxidizer when dry.

Hazardous combustion products

Disodium oxide. Hydrogen chloride. Oxygen. Contact with acids, organic materials, reducing agents or chlorine donors will produce chlorine dioxide gas and heat. Ventilate area with large amounts of air to keep the chlorine dioxide concentration low.

### 6. Accidental Release Measures

Personal precautions, protective equipment and emergency procedures Immediately evacuate personnel to safe areas. Keep unnecessary personnel away. Keep people away from and upwind of spill/leak. Wear appropriate protective equipment and clothing during clean-up. Do not touch damaged containers or spilled material unless wearing appropriate protective clothing. Local authorities should be advised if significant spillages cannot be contained. For personal protection, see section 8 of the SDS. DO NOT USE RAGS, SAWDUST OR OTHER COMBUSTIBLE ABSORBENTS.

Methods and materials for containment and cleaning up Ventilate the contaminated area. Eliminate all ignition sources (no smoking, flares, sparks, or flames in immediate area). Stop the flow of material, if this is without risk. Keep combustibles (wood, paper, oil, etc.) away from spilled material. Absorb in dry sand or earth and place into containers. Use water spray to reduce vapors or divert vapor cloud drift. Do not let the product dry.



Small Spills: Absorb spill with dry sand, earth or other inert material. Neutralize the spilled material before disposal.

Large Spills: Stop the leak, if this is without risk. Dike the spilled material, where this is possible. Absorb in dry sand or earth and place into containers. If not recoverable, dilute with water or flush to holding area and neutralize. Use water spray to reduce vapors or divert vapor cloud drift. Prevent entry into waterways, sewer, basements or confined areas. Never return spills to original containers for re-use. For waste disposal, see section 13 of the SDS.

**Environmental** precautions

Avoid release to the environment. Avoid discharge into drains, water courses or onto the ground. Contact local authorities in case of spillage to drain/aquatic environment.

## 7. Handling and Storage

**Precautions for safe** handling

Use only in a well-ventilated area. Wear chemically resistant protective equipment during handling. Do not breathe mist or vapor. Do not taste or swallow. Keep away from heat. Do not handle, store or open near an open flame, sources of heat or sources of ignition. Protect material from direct sunlight. Do not let the product dry. When using, do not eat, drink or smoke. Keep away from clothing and other combustible materials. Observe good industrial hygiene practices. Avoid release to the environment.

**Conditions for safe** storage, including any incompatibilities

Store in a cool, dry place out of direct sunlight. Store in a well-ventilated place. Storage area should be clearly identified, clear of obstruction and accessible only to trained and authorized personnel. Store away from incompatible materials (see Section 10 of the SDS). Store in original tightly closed container. Do not store near combustible materials. Do not handle or store near an open flame, heat or other sources of ignition.

# 8. Exposure Controls/Personal Protection

**Occupational Exposure Limits** No exposure limits noted for ingredient(s).

**Biological limit values** No biological exposure limits noted for the ingredient(s).

Appropriate engineering controls

Good general ventilation (typically 10 air changes per hour) should be used. Ventilation rates should be matched to conditions. If applicable, use process enclosures, local exhaust ventilation, or other engineering controls to maintain airborne levels below recommended exposure limits. If exposure limits have not been established, maintain airborne levels to an acceptable level. Use separate, corrosion-resistant ventilation system to capture mist or fume. Do not use wood or other combustibles to construct vent system. Prevent entry into bearings or gear boxes, which could cause an explosion. Provide eyewash station.

Individual protection measures, such as personal protective equipment



### Eye/face protection

Wear safety glasses with side shields (or goggles) and a face shield. Provide an emergency eye wash fountain and quick drench shower in the immediate work area.

## Skin protection Hand protection

Gloves impervious to the material are recommended, such as butyl rubber or neoprene gloves. Advice should be sought from glove suppliers.

Other

Where contact is likely, wear chemical-resistant gloves, a chemical suit, rubber boots, and chemical safety goggles plus a face shield. Wear chemical protective equipment that is specifically recommended by the manufacturer. It may provide little or no thermal protection. Eye wash facilities and emergency shower must be available when handling this product.

If Sodium Chlorite dries on some types of fire-retardant clothing it is known to cause an exothermic reaction. The reaction has been known to cause burns to skin. Nomex appears to be the only material not to experience this reaction.

### **Respiratory protection**

In case of insufficient ventilation, wear suitable respiratory equipment. A NIOSH/MSHA approved air-purifying respirator with the appropriate chemical cartridges or a positive-pressure, air-supplied respirator may be used to reduce exposure. Use a positive-pressure air-supplied respirator if there is any potential for an uncontrolled release, exposure levels are not known, or any other circumstances where air-purifying respirators may not provide adequate protection. Respirators should be selected based on the form and concentration of contaminants in air, and in accordance with OSHA (29 CFR 1910.134). Seek advice from respiratory protection specialists.

### **Thermal Hazards**

If Sodium Chlorite dries on some types of fire-retardant clothing it is known to cause an exothermic reaction. The reaction has been known to cause burns to skin. Nomex appears to be the only material not to experience this reaction.

# General hygiene considerations

Keep from contact with clothing and other combustible materials. Remove and wash contaminated clothing promptly. Upon completion of work, wash hands before eating, drinking, smoking or use of toilet facilities. When using do not smoke. Always observe good personal hygiene measures, such as washing after handling the material and before eating, drinking, and/or smoking. Routinely wash work clothing and protective equipment to remove contaminants.

## 9. Physical and Chemical Properties

Appearance Aqueous solution.
Physical state Liquid.



Form Liquid.

**Color** Clear water-white to slightly yellow liquid

**Odor** Odorless to slight Chlorine-like or faint bleach-like

Odor threshold Not available Molecular formula NaClO<sub>2</sub>

Molecular weight 90.45

**pH** 13 to 13.5 (Depends on concentration)

**Melting point/Freezing Point** 1.4 to 79°F (-17 to 26°C) (Depends on concentration)

**Initial boiling point and boiling range** 230 to 234°F (110 to 112 °C)

Flash point

Evaporation rate

Flammability (solid, gas)

Not applicable

Not available

Not available

Upper/lower flammability or explosive limits

Flammability limit – lower (%)
Flammability limit – upper (%)
Explosive limit – lower (%)
Explosive limit – upper (%)
Vapor pressure
Vapor density
Relative density
Not available
Not available
Not available
Not available

Solubility (ies)

Solubility (water) Soluble

Partition coefficient (n-octanol/water)Not availableAuto-ignition temperatureNot availableDecomposition temperatureNot availableViscosityNot available

Other information

**Density** 1.21 - 1.39 g/cm<sup>3</sup> (Depends on concentration)

Flammability Not applicable

**Specific gravity** 1.21 - 1.39 (Depends on concentration)

Surface tension Not available

## 10. Stability and Reactivity

**Reactivity** The product is stable and non-reactive under normal conditions of use,

storage and transport. Reacts on mixing with acids to give toxic chlorine dioxide and chlorine gases. Mixtures with combustibles, if allowed to dry out, are easily ignited by heat or friction and burn vigorously or may

explode.

**Chemical stability** Material is stable under normal conditions. Will decompose if heated.

Absorption of atmospheric carbon dioxide may lower the pH of the

solution, which will cause it to slowly decompose.

**Possibility of hazardous** 

reactions

Contact with acids, organic materials, reducing agents and oxidizing agents

will release toxic gases of chlorine and/or chlorine dioxide.



Conditions to Avoid Keep away from heat, sparks and open flame. Keep away from direct

sunlight and contact with incompatible materials. This product may react

with reducing agents.

**Incompatible materials** Combustible material. Acids. Organic compounds. Oxidizing agents. Metals.

Sulfur and Sulfur-containing materials. Ethylene glycol. Ammonia. Amines.

Phosphorus. Reducing agents.

Hazardous In the event of fire, the following can be released: Chlorine, Chlorine

decomposition products Dioxide.

## 11. Toxicological Information

Information on likely routes of exposure

**Inhalation** Toxic if inhaled.

**Skin contact** Causes severe skin burns.

**Eye contact** Causes serious eye damage/irritation.

**Ingestion** Harmful if swallowed.

Delayed and immediate effects and chronic effects from short-term and long-term exposure

Effects of short-term (acute) exposure

Causes serious eye damage, may cause severe irritation and possibly burns. Symptoms may include stinging, tearing, redness, swelling, and blurred vision.

Causes severe skin burns. Symptoms may include redness, edema, drying, defatting and cracking of the skin.

Acute ingestion of large quantities may also cause anemia due to the oxidizing effects of the chemical.

Material is irritating to mucous membranes and upper respiratory tract. Symptoms may include coughing, bloody nose and sneezing. High concentrations can cause lung damage.

May be harmful or fatal if swallowed. Symptoms may include pain, headache, nausea, vomiting, dizziness, drowsiness and other central nervous system effects.

Effects of long-term (chronic) exposure

Prolonged exposure may cause chronic effects. Dermatitis is likely to occur from repeated or prolonged contact. Other symptoms may include methemoglobinemia (causes bluish discoloration of the skin and mucous

membranes). Will irritate and may cause corrosion of the gastrointestinal tract.

Information on toxicological effects

Acute toxicity

Product Species Test Results

Sodium Chlorite Solution 40%



Acute

Inhalation

 $LC_{50}$ Rat 0.58 mg/L (Calculated ate at 40%)

Oral

 $LD_{50} \\$ Rat 413 mg/kg (Calculated ate at 40%)

Product	Test	Test Results	
Sodium Chlorite Solution 37%			
Dermal	OECD Guideline 435, "In Vitro Membrane Barrier Test Method for Skin Corrosion"	17 min (Average breakthrough time calculated at 37%)	
Components	Species	Test Results	
Sodium Chlorite (CAS 7758-19-2)			
Acute			
LC <sub>50</sub>	Rat	0.23 mg/L (Mist)	
Oral			
LD <sub>50</sub>	Rat	165 mg/kg	
* Estimates for product may be based on	additional component data not shown.		
Skin corrosion/irritation	Can cause severe skin b	Can cause severe skin burns.	
Serious eve damage/eve irritatio	<b>n</b> Can cause serious eve o	Can cause serious eve damage/irritation.	

S

Serious eye damage/eye irritation Can cause serious eye damage/irritation.

Respiratory or skin sensitization

**Respiratory sensitization** Not expected to be a respiratory sensitizer.

Skin sensitizer Not sensitizing.

Germ cell mutagenicity Not expected to be mutagenic.

Carcinogenicity This product is not considered to be a carcinogen by IARC,

ACGIH, NTP, or OSHA.

IARC Monographs. Overall Evaluation of

Carcinogenicity

Sodium Chlorite (CAS 7758-19-2) Not classifiable as to

carcinogenicity to humans.

**OSHA Specifically Regulated** 

Substances (29 CFR 1910.1001-1050)

Not listed.

**Reproductive toxicity** Not classified as a reproductive toxin.



Specific target organ toxicity - single

exposure

Not classified as a specific target organ toxicity -single

exposure.

Specific target organ toxicity - repeated

exposure

Specific Target Organ Toxicity (STOT), Repeated Exposure:

blood, kidneys, liver, spleen.

**Aspiration toxicity** Not expected to be an aspiration hazard.

**Chronic effects** Chronic skin contact with low concentrations may cause

dermatitis. Prolonged or repeated overexposure may cause

blood, liver, spleen and kidney effects.

## 12. Ecological Information

**Ecotoxicity** 

Toxic to aquatic life. In water and soil, sodium chlorite will eventually degrade to

sodium chloride.

**Product Species Test Results** Sodium Chlorite (CAS 7758-19-2) Aquatic Acute Algae  $EC_{50}$ Green algae (Selenastrum capricornutum) 1.2 mg/l Crustacea  $EC_{50}$ Water flea (Daphnia) 0.025 mg/l Fish  $LC_{50}$ Sheepshead minnow (Cyprinodon variegatus) 110 mg/l Chronic Algae  $EC_{50}$ Green algae (Selenastrum capricornutum) 1 mg/l

**Persistence and degradability** Biodegradation is not applicable to inorganic substances.

**Bioaccumulative potential** The product itself has not been tested

**Mobility in soil** In soil, will degrade to sodium chloride but may form chlorine

dioxide in contact with acidic soils. Chlorate is an intermediate product of decomposition; it will slowly degrade to chloride.

Other adverse effects No other adverse environmental effects (e.g. ozone depletion,

photochemical ozone creation potential, endocrine disruption, global

warming potential) are expected from this component.

## 13. Disposal Considerations

**Disposal instructions** 

Collect and reclaim or dispose in sealed containers at licensed waste disposal site. This material and its container must be disposed of as hazardous waste. Do not allow this material to drain into sewers/water supplies. Do not contaminate ponds, waterways or ditches with chemical



or used container. Dispose of contents/container in accordance with

local/regional/national/international regulations.

**Local disposal** regulations

Dispose in accordance with all applicable regulations.

Hazardous waste code The waste code should be assigned in discussion between the user, the

producer and the waste disposal company.

Waste from residues / unused products

Dispose of in accordance with local regulations. Empty containers or liners may retain some product residues. This material and its container must be disposed of in a safe manner (see: Disposal instructions).

**Contaminated packaging** Empty containers should be taken to an approved waste handling site for

recycling or disposal. Since emptied containers may retain product

residue, follow label warnings even after container is emptied.

# **14.Transport Information**

DOT

**UN** number UN 1908

**UN proper shipping name** Chlorite, solution

Transport hazard class(es)

Class 8 **Subsidiary risk** Label(s) 8 Ш Packing group

**Environmental hazards** 

Marine pollutant Yes

Special precautions for user Read safety instructions, SDS and emergency

procedures before handling.

**Special provisions** A3, A6, A7, B2, IB2, N34, T7, TP2, TP24

**Packaging exceptions** None Packaging non bulk 202 **Packaging bulk** 242

**IATA** 

**UN** number UN 1908

**CHLORITE SOLUTION UN proper shipping name** 

Transport hazard class(es)

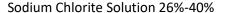
Class 8 **Subsidiary risk Packing group** Ш **Environmental hazards** No 8L

Special precautions for user Read safety instructions, SDS and emergency

procedures before handling.

Other information

**ERG Code** 





Passenger and cargo aircraft Cargo aircraft only

Allowed. Allowed.

**IMDG** 

**UN** number

UN proper shipping name Transport hazard class(es)

> Class Subsidiary risk

Packing group

**Environmental hazards** 

Marine pollutant

**EmS** 

**Special precautions for user** 

UN1908

**CHLORITE SOLUTION** 

Yes

8

Ш

F-A, S-B

Read safety instructions, SDS and emergency

procedures before handling.

Not Available.

Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code DOT



IATA; IMDG



Marine pollutant





**General information** 

DOT Regulated Marine Pollutant. IMDG Regulated Marine Pollutant.

## 15. Regulatory Information

**US federal regulations** This product is a "Hazardous Chemical" as defined by the

> OSHA Hazard Communication Standard, 29 CFR 1910.1200. All components are on the U.S. EPA TSCA

Inventory List. Not regulated.

TSCA Section 12(b) Export

Notification (40 CFR 707, Subpt. D)

**CERCLA Hazardous Substance List** 

(40 CFR 302.4)

Not listed.

Not listed.

**SARA 304 Emergency release** 

notification

Not regulated.

**OSHA Specifically Regulated** 

Substances (29 CFR 1910.1001-1050)

Superfund Amendments and Reauthorization Act of 1986 (SARA)

**Hazard categories** Immediate Hazard - Yes

> Delayed Hazard - Yes Fire Hazard - Yes Pressure Hazard - No Reactivity Hazard - No

**SARA 302 Extremely hazardous** 

Not listed.

substance

SARA 311/312 Hazardous chemical

Nο

**SARA 313 (TRI reporting)** 

Not regulated.

Other federal regulations

Clean Air Act (CAA) Section 112 Not regulated.

**Hazardous Air Pollutants (HAPs) List** 

Clean Air Act (CAA) Section 112(r) Not regulated.

**Accidental Release Prevention (40** 

CFR 68.130)

Safe Drinking Water Act (SDWA) Not regulated.

**US state regulations** 

**US. California Controlled Substances.** Not listed.

**CA Department of Justice (California** 



**Health and Safety Code Section** 

11100)

**US. Massachusetts RTK - Substance** Sodium Chlorite (CAS 7758-19-2)

List

**US. New Jersey Worker and** Sodium Chlorite (CAS 7758-19-2)

Community Right-to-Know Act
US Pennsylvania RTK – Hazardou

**US. Pennsylvania RTK – Hazardous** Sodium Chlorite (CAS 7758-19-2)

Substances

US. Rhode Island RTK Not regulated.

**US. California Proposition 65** California Safe Drinking Water and Toxic Enforcement

Act of 1986 (Proposition 65): This material is not known to contain any chemicals currently listed as carcinogens

or reproductive toxins.

#### **International Inventories**

Country(s) or region	Inventory name	On inventory (yes/no)*
Australia	Australian Inventory of Chemical Substances (AICS)	Yes
Canada	Domestic Substances List (DSL)	Yes
Canada	Non-Domestic Substances List (NDSL)	No
China	Inventory of Existing Chemical Substances in China (IECSC)	Yes
Europe	European Inventory of Existing Commercial Chemical Substances (EINECS)	Yes
Europe	European List of Notified Chemical Substances (ELINCS)	No
Japan	Inventory of Existing and New Chemical Substances (ENCS)	Yes
Korea	Existing Chemicals List (ECL)	Yes
New Zealand	New Zealand Inventory	Yes
Philippines	Philippine Inventory of Chemicals and Chemical Substances (PICCS)	Yes
United States & Puerto Rico	Toxic Substances Control Act (TSCA) Inventory	Yes

<sup>\*</sup>A "Yes" indicates that all components of this product comply with the inventory requirements administered by the governing country(s).

A "No" indicates that one or more components of the product are not listed or exempt from listing on the inventory administered by the governing country(s).

## **16.Other Information**

**Issue date** 3/31/2022

Revision # 5

**Revision Indicator** Clarified precautionary statements, added FR clothing precaution and

address updated.

**List of abbreviations** ACGIH: American Conference of Governmental Industrial Hygienists

**CAS: Chemical Abstract Services** 



CERCLA: Comprehensive Environmental Response, Compensation and

Liability Act of 1980

CFR: Code of Federal Regulations DOT: Department of Transportation EPA: Environmental Protection Agency

EPCRA: Emergency Planning and Community Right-to-Know Act

ERG: Emergency Response Guidebook HSDB® - Hazardous Substances Data Bank

IARC: International Agency for Research on Cancer IATA: International Air Transport Association

IBC: Intermediate Bulk Container

IDLH: immediately dangerous to life or health IMDG: International Maritime Dangerous Goods

LC: Lethal Concentration

LD: Lethal Dose

NIOSH: National Institute of Occupational Safety and Health

NOEC: No observable effect concentration

NTP: National Toxicology Program

OECD: Organization for Economic Cooperation and Development

OEL: National occupational exposure limits

OSHA: Occupational Safety and Health Administration

PEL: Permissible exposure limit

RCRA: Resource Conservation and Recovery Act

**RQ: Reportable Quantity** 

RTECS: Registry of Toxic Effects of Chemical Substances

SAR: supplied-air respirator

SCBA: self-contained breathing apparatus

SDS: Safety Data Sheet

STEL: Short Term Exposure Limit TWA: Time Weighted Average

**UN: United Nations** 

References None.

#### Disclaimer

Information presented in this SDS is furnished in accordance with OSHA's Hazard Communication Standard (HCS) 2012.

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